

## Formula Sheet

$$\ln n! \approx n \ln n - n \quad \sum_{n=0}^{\infty} r^n = \frac{1}{1-r} \quad (1)$$

$$\int_{-\infty}^{\infty} e^{-\alpha x^2} dx = \left(\frac{\pi}{\alpha}\right)^{1/2} \quad \int_{-\infty}^{\infty} e^{-\alpha x^2} x^2 dx = \frac{\sqrt{\pi}}{2} \alpha^{-3/2} \quad (2)$$

### Distributions

$$\text{Binomial distribution} \quad P(n, N) = \frac{N!}{n!(N-n)!} p^n q^{N-n} \quad (3)$$

$$\text{Gaussian Distribution} \quad P(x)dx = \frac{1}{\sqrt{2\pi}\sigma} e^{-(x-\mu)^2/2\sigma^2} dx \quad (4)$$

$$\text{Poisson Distribution} \quad P(n) = \frac{\mu^n}{n!} e^{-\mu} \quad \mu = Np \quad (5)$$

### Thermodynamics

$$dW = \bar{p}dV \quad \Delta \bar{E} = Q - W + \mu \Delta \bar{N} \quad (6)$$

$$\beta = \frac{1}{k_B T} = \frac{\partial \ln \Omega(E)}{\partial E} \quad S \equiv \text{entropy} \equiv k_B \ln \Omega \quad (7)$$

$$\frac{1}{T} = \left. \frac{\partial S}{\partial E} \right|_V \quad (8)$$

$$\bar{X}_\alpha \equiv -\frac{\overline{\partial E_r}}{\partial x_\alpha} \quad \frac{\partial S}{\partial x_\alpha} = \frac{\bar{X}_\alpha}{T} \quad (9)$$

$$dQ = TdS = d\bar{E} + dW - \mu d\bar{N} \quad dW = \sum_{\alpha=1}^n \bar{X}_\alpha dx_\alpha \quad (10)$$

$$\Delta S \geq 0 \quad \text{For an isolated system} \quad (11)$$

### Numbers

$$T_C = T_K - 273.15 \quad \text{degrees Celsius} \quad (12)$$

$$R = (8.3143 \pm 0.0012) \text{ joules mole}^{-1} \text{ deg}^{-1} \quad (13)$$

$$k_B = (1.38054 \pm 0.00018) \times 10^{-16} \text{ ergs degree}^{-1} \quad (14)$$

$$1 \text{ calorie} = 4.1840 \text{ joules} \quad 1 \text{ Joule} = 10^7 \text{ ergs} \quad (15)$$

$$N_a = 6.023 \times 10^{23} \frac{\text{molecules}}{\text{mole}} \quad (16)$$

$$1 \text{ atm} = 1.01325 \times 10^6 \text{ dynes/cm}^2 = 1.013 \times 10^5 \text{ N/m}^2 \quad (17)$$

$$C_y = \left. \frac{dQ}{dT} \right|_y = T \left. \frac{\partial S}{\partial T} \right|_y \quad (18)$$

$$\Delta S = S(T_f) - S(T_i) = \int_{T_i}^{T_f} \frac{C_V(T')}{T'} dT' \quad \text{For fixed volume} \quad (19)$$

$$\bar{\epsilon}_i = \frac{1}{2} k_B T \quad \text{Equipartition Theorem} \quad (20)$$

### Ideal Gas

$$pV = Nk_B T = \nu RT \quad c_p = c_V + R \quad \bar{E} = \frac{3}{2} Nk_B T \quad (21)$$

$$Z = \frac{Z'}{N!} = \frac{\zeta^N}{N!} \quad S = k_B N \left[ \ln \frac{V}{N} + \frac{3}{2} \ln T + \sigma_o \right] \quad (22)$$

$$\sigma_o = \sigma + 1 = \frac{3}{2} \ln \left( \frac{2\pi m k_B}{h^2} \right) + \frac{5}{2} \quad (23)$$

$$pV^\gamma = \text{constant} \quad \gamma = \frac{c_p}{c_V} \quad V^{\gamma-1} T = \text{constant} \quad (24)$$

### Maxwell Relations and Thermodynamic Functions

$$\left( \frac{\partial T}{\partial V} \right)_S = - \left( \frac{\partial p}{\partial S} \right)_V \quad \left( \frac{\partial T}{\partial p} \right)_S = \left( \frac{\partial V}{\partial S} \right)_p \quad (25)$$

$$\left( \frac{\partial S}{\partial V} \right)_T = \left( \frac{\partial p}{\partial T} \right)_V \quad - \left( \frac{\partial S}{\partial p} \right)_T = \left( \frac{\partial V}{\partial T} \right)_p \quad (26)$$

$$\text{Energy } E = E(S, V) \quad dE = TdS - pdV \quad (27)$$

$$\text{Enthalpy } H = H(S, p) = E + pV \quad dH = TdS + Vdp \quad (28)$$

$$\text{Helmholtz Free Energy } F = F(T, V) = E - TS \quad dF = -SdT - pdV \quad (29)$$

$$\text{Gibbs Free Energy } G = G(T, p) = E - TS + pV \quad dG = -SdT + Vdp \quad (30)$$

$$T = \left( \frac{\partial E}{\partial S} \right)_V = \left( \frac{\partial H}{\partial S} \right)_p \quad p = - \left( \frac{\partial E}{\partial V} \right)_S = - \left( \frac{\partial F}{\partial V} \right)_T \quad (31)$$

$$V = \left( \frac{\partial H}{\partial p} \right)_S = \left( \frac{\partial G}{\partial p} \right)_T \quad S = - \left( \frac{\partial F}{\partial T} \right)_V = - \left( \frac{\partial G}{\partial T} \right)_p \quad (32)$$

$$\text{Clausius - Clapeyron Equation } \frac{dp}{dT} = \frac{\Delta s}{\Delta v} \quad (33)$$

### Engines and Refrigerators

$$q_1 = w + q_2 \quad \frac{q_2}{q_1} \leq \frac{T_2}{T_1} \quad (34)$$

$$\text{Efficiency : } \quad \eta \equiv \frac{w}{q_1} = \frac{q_1 - q_2}{q_1} < 1 \quad \eta_{max} = \frac{T_1 - T_2}{T_1} \quad (35)$$

### Statistical Mechanics

$$\bar{A} = \frac{1}{Z} \sum_R A_R e^{-\beta E_R} \quad Z = \sum_r e^{-\beta E_r} \quad (36)$$

$$\bar{E} = -\frac{\partial \ln Z}{\partial \beta} \quad S = k_B (\ln Z + \beta \bar{E}) \quad \bar{X} = \frac{1}{\beta} \frac{\partial \ln Z}{\partial x} \quad (37)$$

$$\bar{p} = \frac{1}{\beta} \frac{\partial \ln Z}{\partial V} \quad \bar{M} = \frac{1}{\beta} \frac{\partial \ln Z}{\partial H} \quad (38)$$

$$F = -k_B T \ln Z \quad C_V = k_B \beta^2 \frac{\partial^2 \ln Z}{\partial \beta^2} = -T \left. \frac{\partial^2 F}{\partial T^2} \right|_V \quad (39)$$

$$\mathcal{Z} = \sum_r e^{-\beta(E_r - \mu N_r)} \quad \mu = -\frac{\partial E'}{\partial N'} \quad (40)$$

$$\bar{N} = \frac{1}{\beta} \frac{\partial \ln \mathcal{Z}}{\partial \mu} \quad \bar{E} = \mu \bar{N} - \frac{\partial}{\partial \beta} \ln \mathcal{Z} \quad (41)$$

$$\ln Z = -\beta \mu N + \ln \mathcal{Z} \quad \mu = -\frac{1}{\beta} \frac{\partial \ln Z(N)}{\partial N} = \frac{\partial F}{\partial N} \quad (42)$$

$$f(\vec{v}) d^3 r d^3 v = n \left( \frac{m}{2\pi kT} \right)^{3/2} e^{-mv^2/2kT} d^3 r d^3 v \quad \text{Maxwell velocity distribution} \quad (43)$$

$$\bar{n}_s = -\frac{1}{\beta} \frac{\partial \ln Z}{\partial \epsilon_s} \quad \bar{N} = \sum_s \bar{n}_s \quad (44)$$

$$\bar{n}_s = N \frac{e^{-\beta \epsilon_s}}{\sum_r e^{-\beta \epsilon_r}} \quad \text{Maxwell Boltzmann Distribution} \quad (45)$$

$$\bar{n}_s = \frac{1}{e^{\beta \epsilon_s} - 1} \quad \text{Planck distribution} \quad (46)$$

$$\bar{n}_s = \frac{1}{e^{\beta(\epsilon_s - \mu)} - 1} \quad \text{bosons} \quad \bar{n}_s = \frac{1}{e^{\beta(\epsilon_s - \mu)} + 1} \quad \text{fermions} \quad (47)$$

$$\text{Black Body Radiation : } \quad \bar{u}(\omega, T) d\omega = \frac{\hbar \omega^3}{\pi^2 c^3} \frac{d\omega}{(e^{\beta \hbar \omega} - 1)} \quad \bar{u}_o(T) = \frac{\pi^2 (k_B T)^4}{15 (\hbar c)^3} \quad (48)$$

$$\mathcal{P}_e^{tot} = \frac{1}{4} c \bar{u}_o = \sigma T^4 \quad \sigma \equiv \frac{\pi^2}{60} \frac{k_B^4}{c^2 \hbar^3} \quad (49)$$

$$\text{Free Electron Gas : } \quad k_F = \left( 3\pi^2 \frac{N}{V} \right)^{1/3} \quad C_V = \frac{\pi^2}{2} R \cdot \frac{T}{T_F} \quad (50)$$

$$\text{Einstein Heat Capacity : } \quad C_V = 3Nk_B \left( \frac{\theta_E}{T} \right)^2 \frac{e^{\theta_E/T}}{(e^{\theta_E/T} - 1)^2} \quad (51)$$

$$\text{Debye Heat Capacity : } C_V = \frac{12\pi^4}{5} N k_B \left( \frac{T}{\Theta_D} \right)^3 \quad (52)$$

$$\text{Bose Einstein Condensation : } n_1 = N \left[ 1 - \left( \frac{T}{T_C} \right)^{3/2} \right] \quad \mu = -k_B T \ln \left( \frac{1}{n_1} + 1 \right) \quad (53)$$

### Spin Systems

$$\vec{\mu} = g\mu_o \mathbf{S} \quad (54)$$

$$\mathcal{H}_o = -g\mu_o \sum_{j=1}^N \mathbf{S}_j \cdot \mathbf{H}_o \quad (55)$$

$$\text{Heisenberg Hamiltonian : } \mathcal{H} = -2J \sum_{i>j} \mathbf{S}_i \cdot \mathbf{S}_j \quad (56)$$

If we take  $J_{ij} = J$ , then

$$\mathcal{H} = -2J \sum_{i>j} \mathbf{S}_i \cdot \mathbf{S}_j \quad (57)$$

$$\text{Ising Model : } \mathcal{H} = -2J \sum_{i>j} S_{iz} S_{jz} \quad (58)$$